

UV absorption spectroscopy to study the role of CF radicals during cryogenic etching of Si in CF₄ plasmas

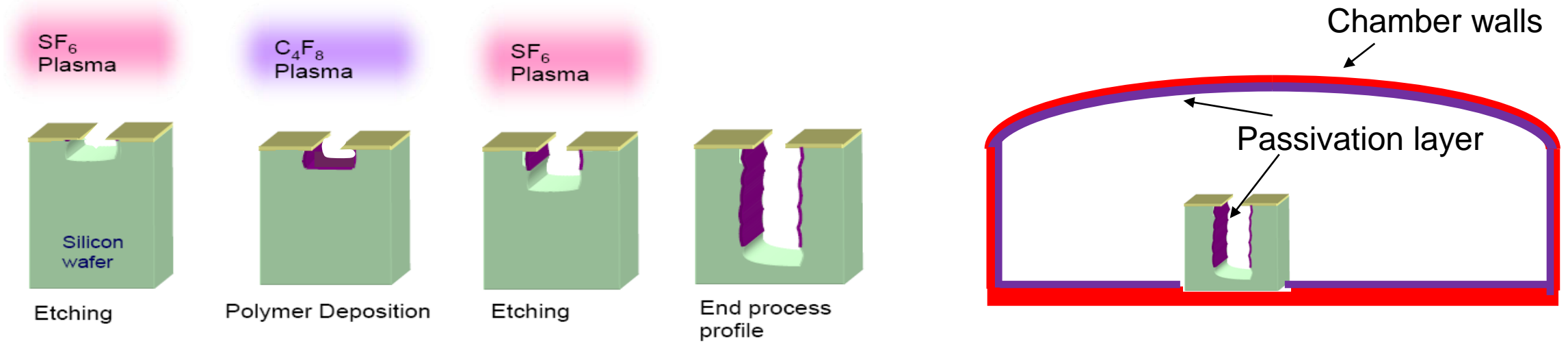
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Context: Bosch process

F chemistry at room temperature – sidewall passivation with gas modulation



Time-multiplexed process for deep silicon etching

- Cycles composed of SF_6 etch plasma followed by C_4F_8 deposition plasma
- Passivation layer = Fluoropolymer (C_xF_y) produced by the decomposition of the injected gas

Advantages

- Process **usually** at **room temperature**
- **Robust** process, repeatable

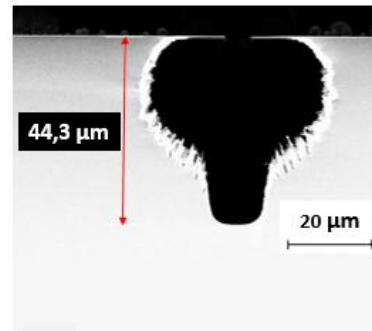
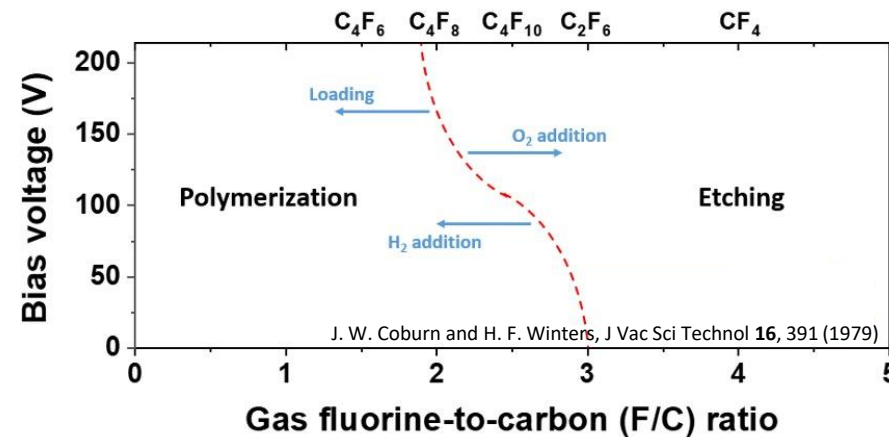
Drawbacks

- « Scalloping » at the sidewalls
- Deposition of fluorocarbon layer on chamber walls -> Chamber cleaning necessary to prevent process drifts
- Global warming potential of C_4F_8

Switching to less polymerizing gas ?

- Low F/C ratio gases are highly polymerizing

⇒ Deposition regime whatever the substrate temperature



+20°C

CF₄

→ Isotropic profile

Nos et al, Appl. Phys. Lett. 126, 031602 (2025)

- High F/C ratio gas-based plasmas lead to an etch regime at room temperature

⇒ Is it possible to promote deposition on feature sidewalls by only decreasing the substrate temperature while keeping chamber walls clean ?

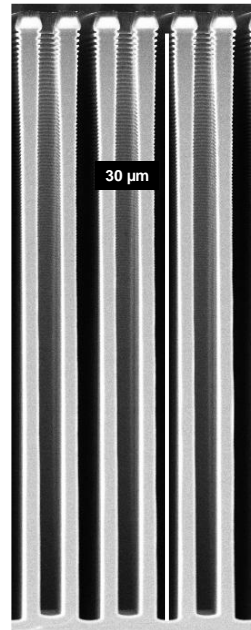
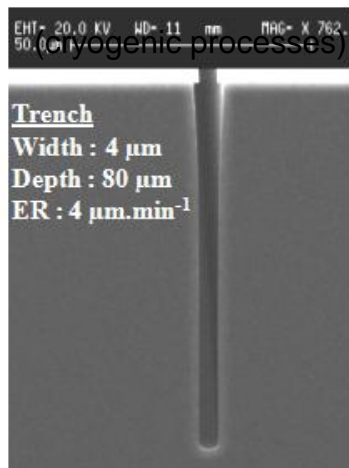
→ Bosch process at cryogenic temperature with CF₄ ?

Cryogenic etching

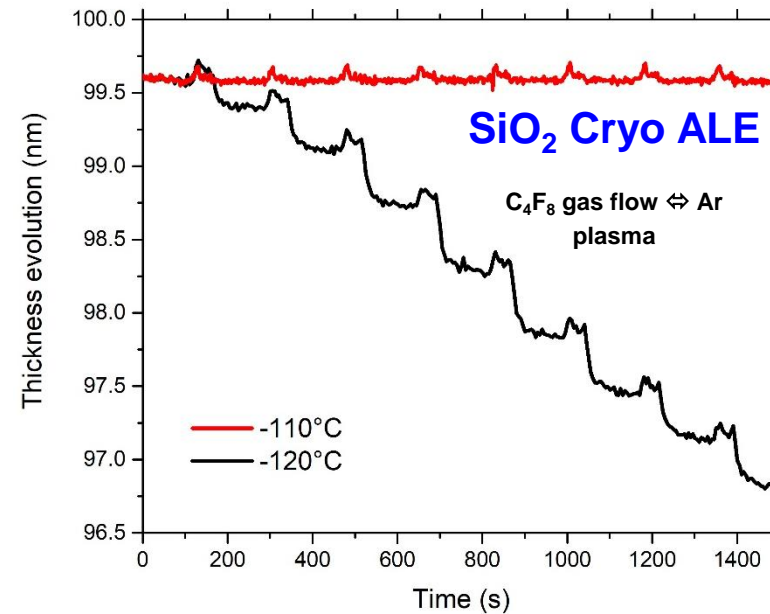
Benefits of substrate cryogenic cooling for plasma etching

- ⇒ Increased surface residence time of species => passivation, capillary condensation, physisorption...
- ⇒ Less plasma-induced damage
- ⇒ Less contamination of chamber walls (layers deposited only on cooled surfaces)

Deep silicon etching



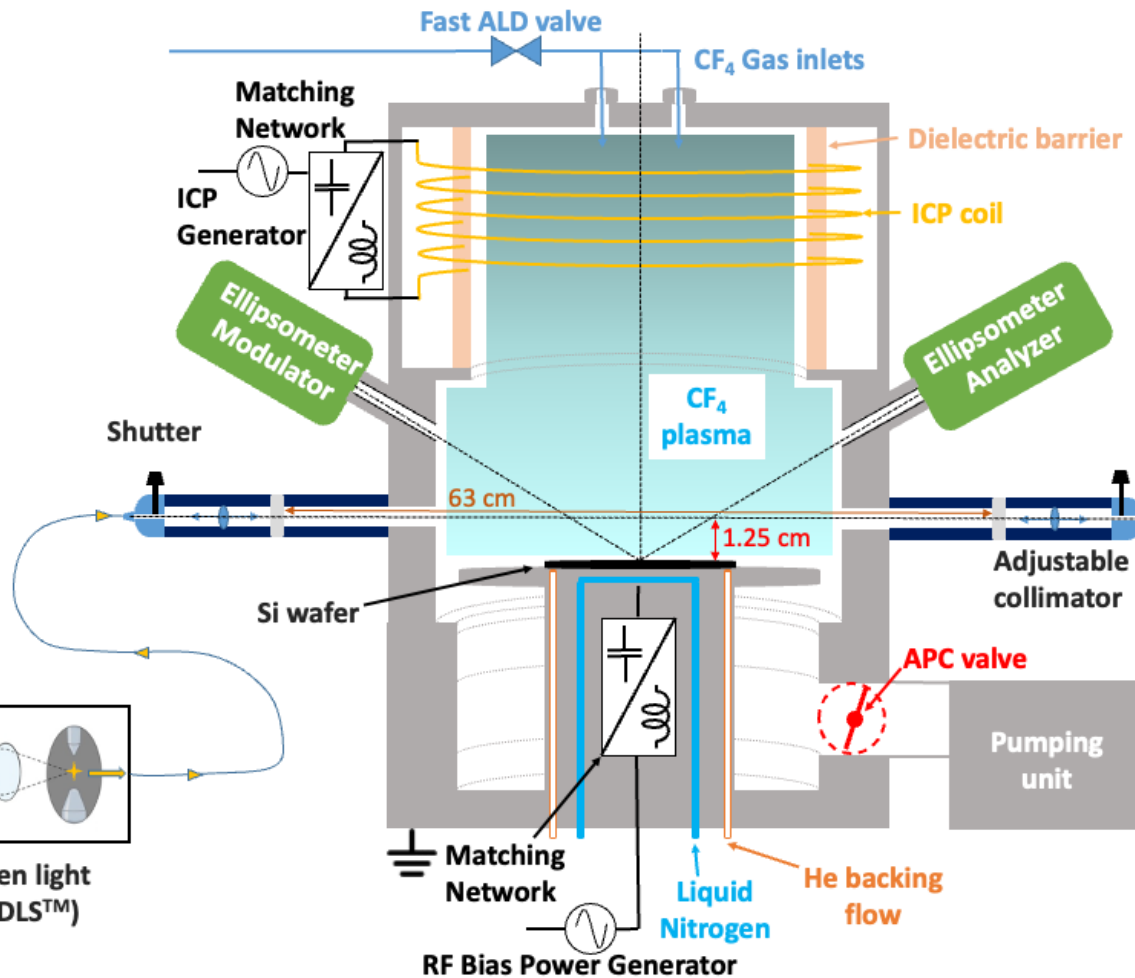
R. Dussart *et al.*, J. Phys. D: Applied Physics, 47 (2014) 123001



G. Antoun *et al.*, Appl. Phys. Lett. 115 (2019) 153109

Experimental setup

Oxford Instruments ICP reactor

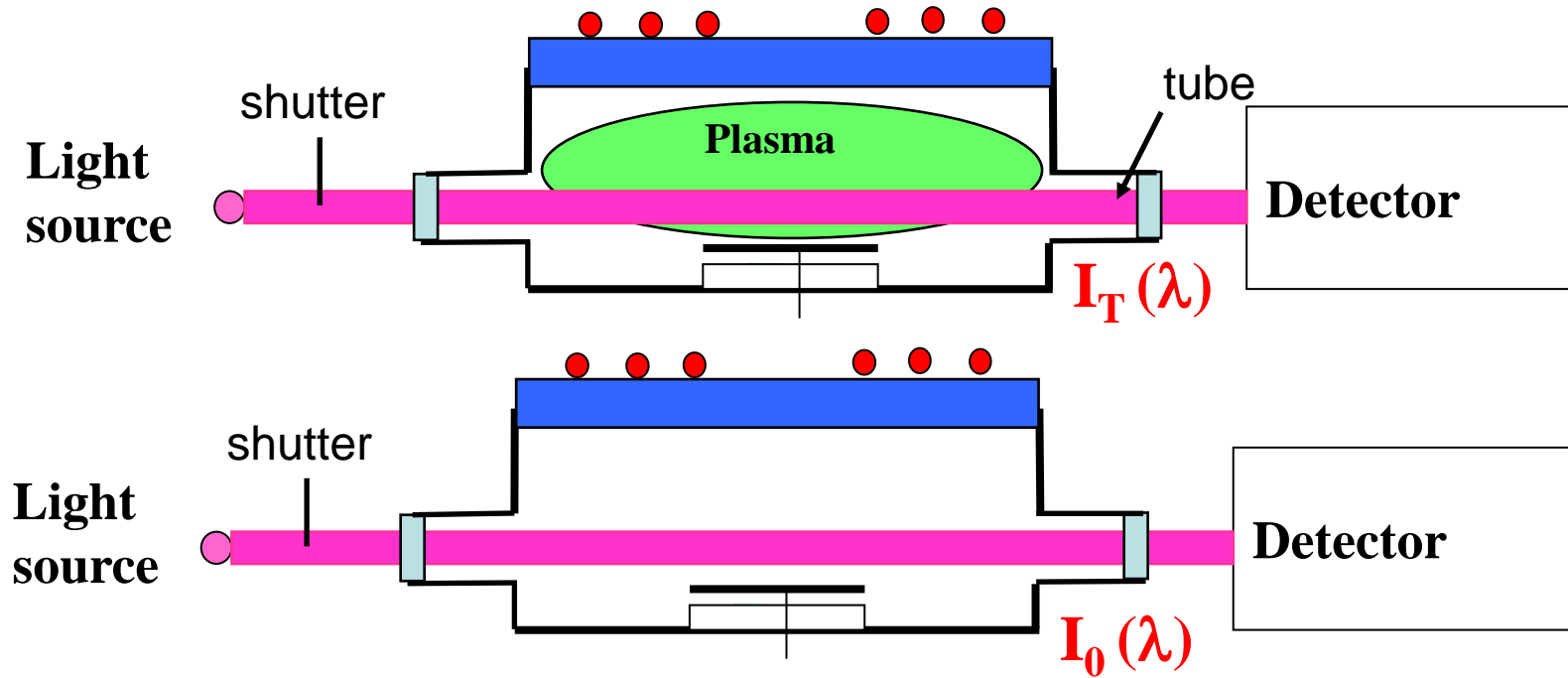


Equipment:

- Hamamatsu Energetiq LDLS
=> 170 – 2500 nm
- Additional tubes on the viewports
=> Connection du optical fiber, collimation, protection of windows
- Princeton Instruments Acton Spectrometer, 750 mm, 3 gratings
- Pylon CCD detector (200 – 1050 nm)

Optical spectrometer
+
CCD detector

Optical absorption spectroscopy



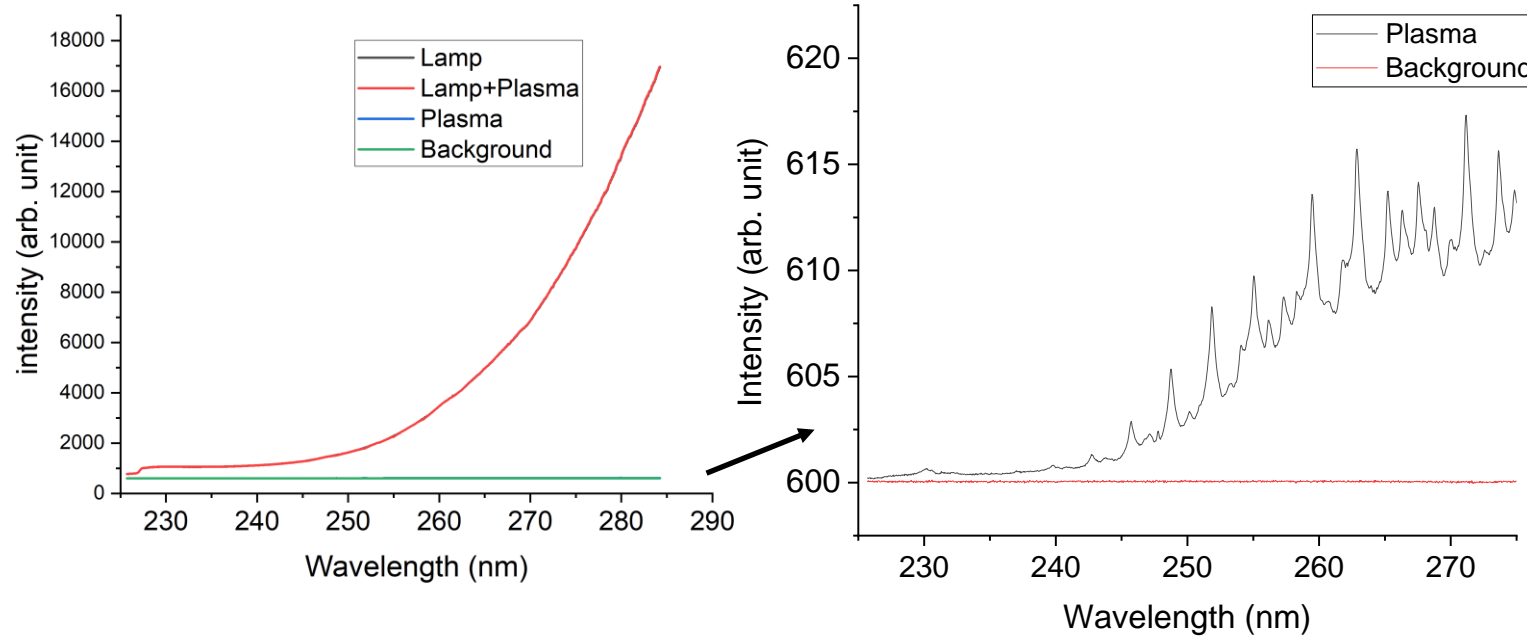
Beer-Lambert law: $I_T(\lambda) = I_0(\lambda) e^{-n \cdot \sigma(\lambda) L}$

By measuring $I_0(\lambda)$ and $I_T(\lambda)$:

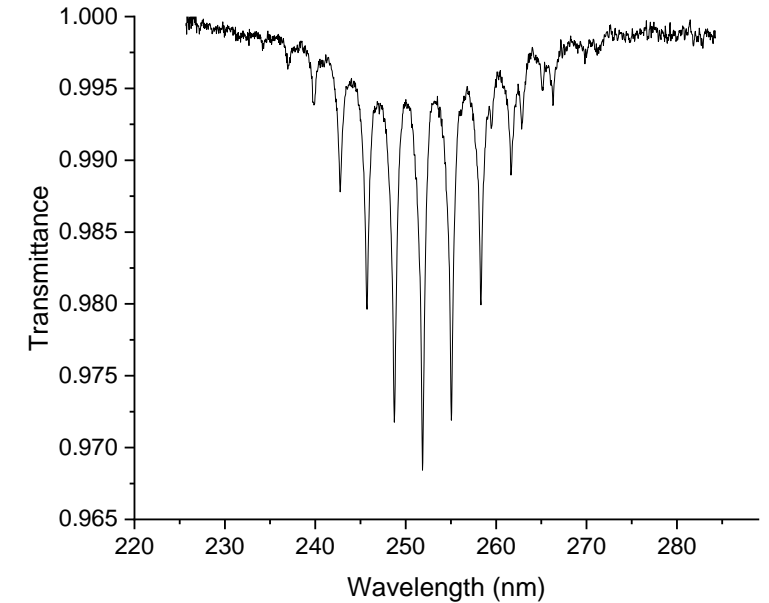
- 1) Shape of $\sigma(\lambda) \Rightarrow$ species temperature
- 2) Absolute concentration of n

\Rightarrow Flux of radicals incident on the substrate: $\Phi = \frac{1}{4} n \sqrt{\frac{8kT}{\pi m}}$

Optical absorption spectroscopy



CF₂ A(0,v',0)-X(0,0,0)
Absorption spectrum



I_{LP} : Lamp on – plasma on, shutter open
 I_P : Lamp off – plasma on, shutter closed
 I_L : Lamp on – plasma off, shutter open
 I_{BG} : Lamp off – plasma off, shutter closed

$$\text{Transmittance} = \left(\frac{I_{LP} - I_P}{I_L - I_{BG}} \right)$$

Thanks to multichannel detector, a large wavelength range can be detected simultaneously

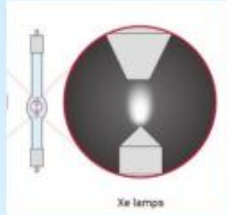
Challenge: light source has to be stable to avoid baseline fluctuation

⇒ High sensitivity needed for some species (10^{-4} absorbance with few sec acquisition)

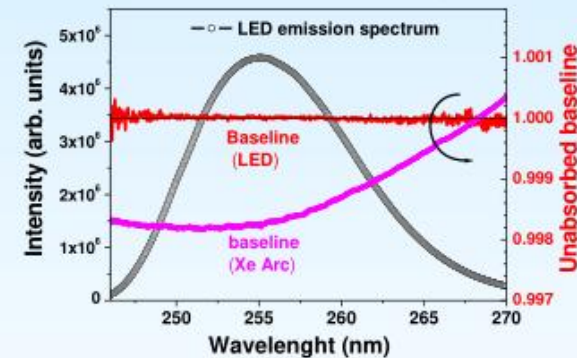
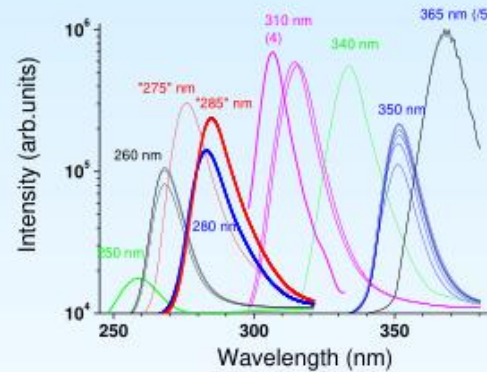
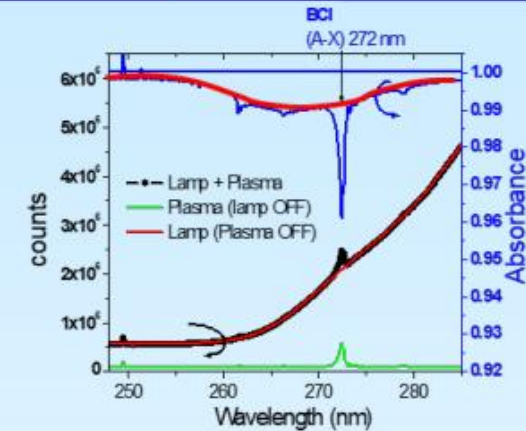
Improvement of lamp source stability over 20 years

➤ Xe arc

Arc = spatial + intensity fluctuations → the unabsorbed baseline $I_0(t1)/I_0(t2)$ oscillates



→ Sensitivity limited by baseline oscillations ($> 10^{-3}$)



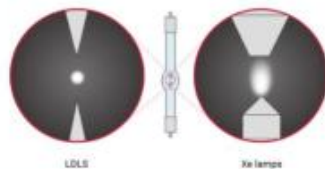
→ Sensitivity is limited by the baseline noise ($< 10^{-4}$) → gain factor 10 in sensitivity...but spectral range of each LED restricted

➤ Laser driven light sources (LDLS)

→ stability almost as good as LED but wider spectral range (180 nm → visible)



High radiance from a small plasma
High radiance emission from a luminous point of 0.1 mm diameter



→ A good choice to detect small polyatomic radicals...

Optical absorption spectroscopy: experimental protocol

Influence of the substrate temperature on the CF and CF₂ radical densities

- Broadband UV absorption spectroscopy: measurement procedure

- x3
- 1) Plasma OFF – Lamp OFF (shutter closed) - Acquisition of background spectrum (I_{BG})
 - 2) Plasma OFF – Lamp ON (shutter open) - Acquisition of lamp emission spectrum (I_L) x2
 - 3) Plasma ON – Lamp OFF (shutter closed) – Acquisition of plasma emission spectrum (I_P) X2
 - 4) Plasma ON – Lamp ON (shutter open) – Acquisition transmission spectrum (I_{LP})

- Experiment performed at **two substrate temperatures**: 20 °C and -130 °C
- Absorption spectra plotted from the average of 3 separate measurements
- Absorbance A is given by Beer Lambert's law:

$$-\ln \left(\frac{I_{LP} - I_P}{I_L - I_{BG}} \right) = -\ln \left(\frac{I_T}{I_0} \right) = A = \sigma \cdot n \cdot L$$

L = absorption path
 σ = absorption cross section
n = species density

J. Luque et al., J Chem Phys **118**, 622 (2003)

Influence of the substrate temperature on gas phase CF and CF₂ radical densities

- CF₄ plasma: 50 sccm, 4 Pa, 1500 W ICP, no bias power

- Clear decrease in the intensity of absorption signal from 20°C to -130°C

- Calculated CF density at 20°C: $1 \times 10^{13} \text{ cm}^{-3}$

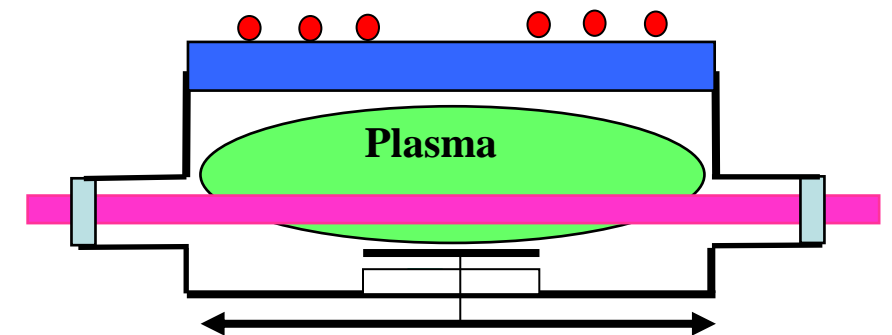
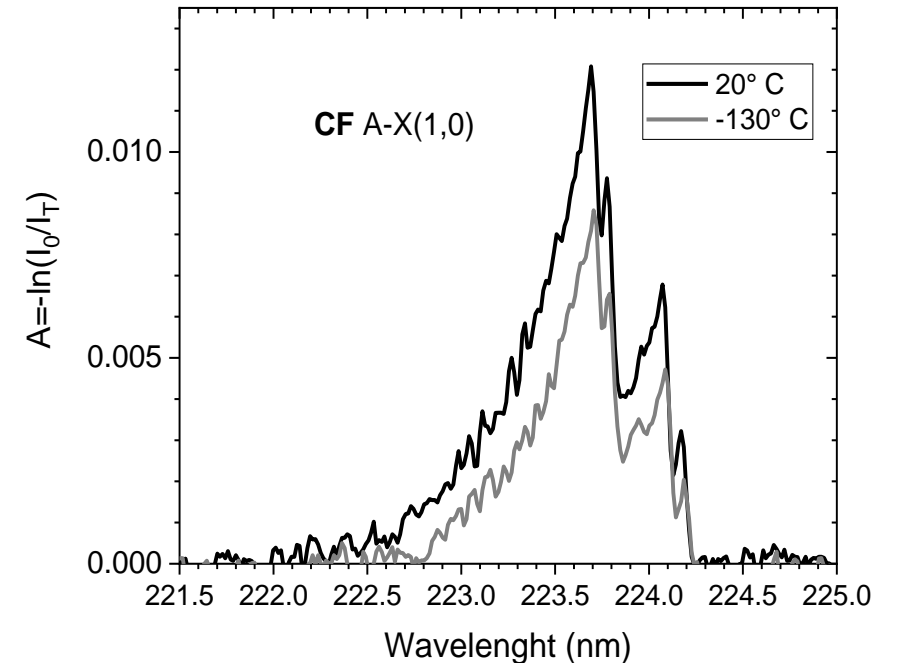
- Calculated CF density at -130°C: $6 \times 10^{12} \text{ cm}^{-3}$

=> Decrease of CF density by 40%

- Cooled wafer area << total chamber wall surface at room T

=> Strong increase of CF loss rate on the surface cooled at -130°C

CF species, A-X(1,0) transition



Absorption path = reactor diameter

Influence of the substrate temperature on gas phase CF and CF₂ radical densities

CF₄ plasma: 50 sccm, 4 Pa, 1500 W ICP, no bias power

- No significant dependence of CF₂ radical density to the substrate temperature (around $1.7 \times 10^{13} \text{ cm}^{-3}$)

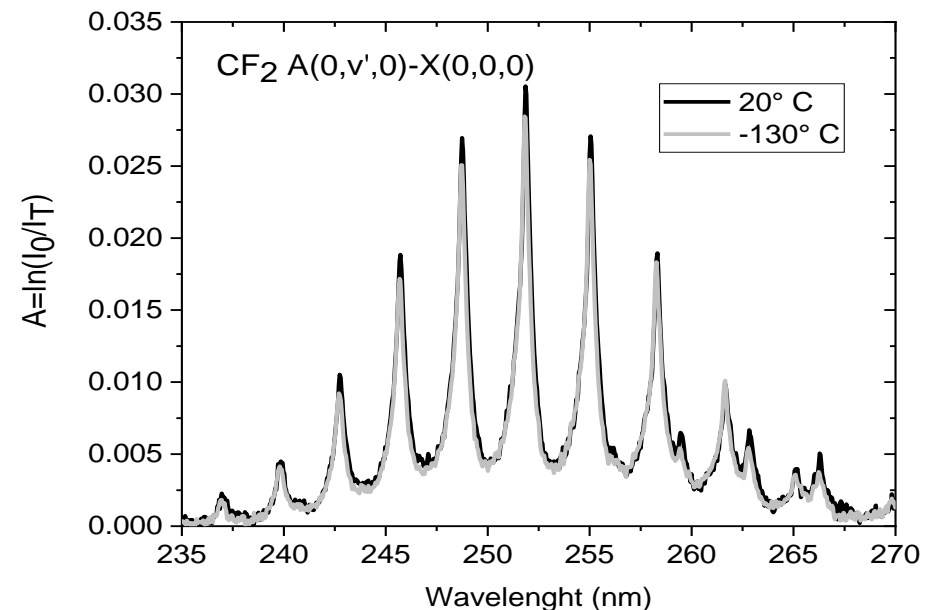
**=> No sticking of CF₂ on surfaces
(usually produced at surfaces)**

→ Absorption path = reactor diameter + tubes

**=> CF expected to be the main precursor to CF_x
deposition at low temperature**

- Competition between etching and deposition in CF₄ plasma turns to deposition at -130°C
- **=> Favored by enhanced sticking of CF radicals on surface cooled at cryogenic temperature**
- By neglecting etching component, sticking probability can be estimated to a minimum of 10%

CF₂ species, A(0,v',0)-X(0,0,0) transition



Cryo-Bosch process with CF_4 passivation

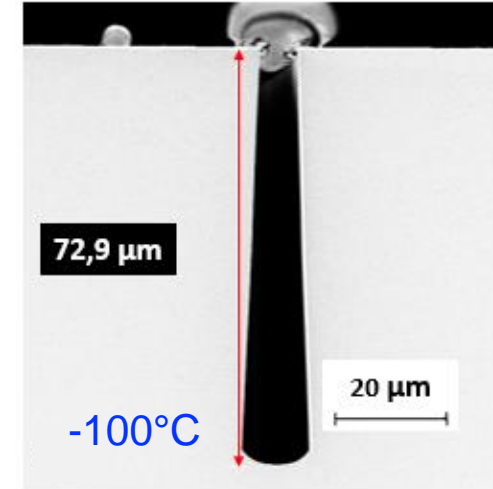
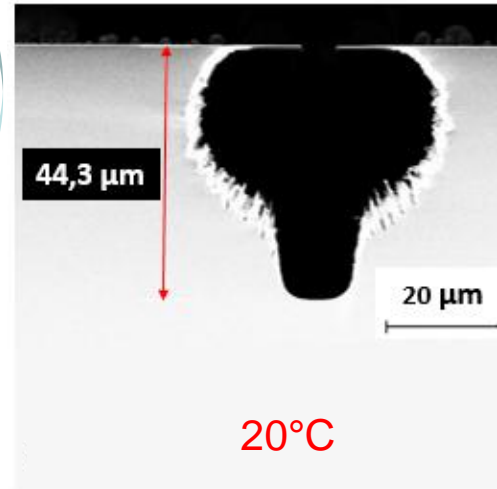
Trenches cross-section view (CD = 6 μm)

x200

Etching step: 3 s
300 sccm SF_6 , 3 Pa, 1500 Ws, -135 Vb

Passivation step: 2 s
20 sccm CF_4 , 1 Pa, 1500 Ws, -65 Vb

- Mostly **isotropic** profile at **20°C**
- **Anisotropic** profile at **-100°C**

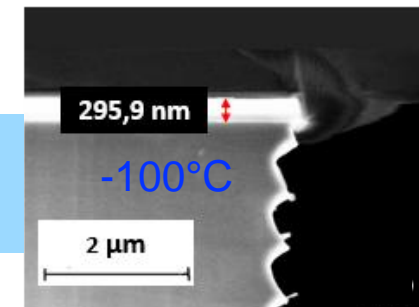


Top view – SiO_2 mask (initial thickness 1.2 μm)

=> Strong CF_x deposition at -100°C

=> Enhanced passivation at -100°C

- Selectivity at **20°C**: 45:1
- Selectivity at **-100°C**: 80:1



Nos et al, *Appl. Phys. Lett.* 126, 031602 (2025)

Conclusions

- **CF₄ plasma:** Transition from Si etching to CF_x deposition by decreasing the substrate temperature from room T to cryogenic T

- ⇒ Significant decrease of CF radical density observed by UV absorption spectroscopy in cryogenic conditions
- ⇒ Significant increase of sticking probability of CF radicals on cold surfaces
- ⇒ CF species are expected to be the main precursor of CF_x layer



- **Bosch process** can be achieved at **cryogenic temperature** by using CF₄ passivation plasma instead of C₄F₈

- => The CF_x layer grows only on cooled surface
- => No deposition on reactor walls at room T
- => Less contamination of chamber walls, less global warming potential

Thank you !

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thank



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